

DOCKET NO.: 208285US90

## IN THE UNITED STATES PATENT AND TRADEMARK OFFICE

IN RE APPLICATION OF:

Naoto IKEGAWA et al : GROUP ART UNIT: 1773

SERIAL NO: 09/871,896 : EXAMINER: Nikolas UHLIR

FILED: JUNE 4, 2001

FOR: LAMINATE

#### DECLARATION UNDER 37 C.F.R. 61.132

COMMISSIONER FOR PATENTS ALEXANDRIA, VIRGINIA 22313

SIR:

Now comes IKEGAWA, Nauto who deposes and states:

- 1. That I am a graduate of of Tech. and received a Doctor degree in the year 1996
- 2. That I have been employed by Works for 13 years as a regular employee in the field of molding technology.
- 3. That the following experiments were carried out by me or under my direct supervision and control.
- a) An aromatic polyamide (poly(phthalamide)) base resin was prepared by adding a filler material of bonic aluminum of an amount of 70% (by mass) thereto.
- b) A liquid crystal polyester base resin was prepared by adding a filler of fibrous potassium titanate at an amount of 50% (by mass) thereto.
- e) A polyether ether ketone was prepared by adding a filler of boric aluminum at an amount of 20% (by mass) thereto.

A sample of each hase resin produced above was treated by nitrogen plasma, oxygen plasma and argon plasma and a copper metal layer was deposited on the base resins using the procedures set forth in the present application at the section titled "Examples" on pages 29-30.

The adhesion between the base resin and the deposited copper was then measured and the results of the experiments are shown in Table 1.

Table 1

					ATOM
Baseresin	Filler		Nitrogeo	Oxygen .	Aigon -
	Material	Configuration	plasma		1.04N/mm
Arpmatic	botic	70%	1,1N/mm	0.77N/mm	, ,
polysmide (poly(phthalamide))	aluminum (dismeter	,			
	0.5-1,0µm,				
	length 10- 30,400)			,	
Thought amortal	fibrous	50%	0.55N/mm	0.25N/mm	0.37N/mm
Liquid crystal polyester	potaesium titanete			•	
	(dismeter 0.3-0.6µm,				
	length 10- 20µm)				<u> </u>
Polyether ether ketone	poric	20%	0.56N/mm		,
	(diameter 0.5-1.0µm.				
• • •	length 10- 30µm)_				

4. The results of the experiments set forth in Table 1 demonstrate for each different type of base resin containing a different amount of filler material in a different amount within the range of the present claims a higher adhesion for deposited metal to base resin treated by mirrogen plasma over deposited metal to a base resin treated by oxygen plasma or argon plasma from a range of approximately 6% greater adhesion up to 220% greater adhesion. Therefore, it is clear that nitrogen plasma treatment of a base resin containing filler material.

according to Claim 1 produces superior adhesion between the base resin and deposited metal, as compared to adhesion between a base resin with filler material treated by oxygen plasma or argon plasma.

- 5. The undersigned petitioner declares further that all statements made herein of his own knowledge are true and that all statements made on information and belief are believed to be true; and further that these statements were made with the knowledge that willful false statements and the like so made are punishable by fine or imprisonment, or both, under Section 1001 of Title 18 of the United States Code and that such willful false statements may jeopardize the validity of this application or any patent issuing therefrom.
  - 6. Further deponent saith not

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Signature IKEGAWA, Naoto
Deta



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Naoto IKEGAWA et 21

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### DECLARATION UNDER 37 C.F.R. 61.132

COMMISSIONER FOR PATENTS ALEXANDRIA, VIRGINIA 22313

SIR:

Now comes NAKATA, Kimiaki who deposes and states:

- 1. That I am a graduate of Yamoguchi and received a Master degree in University
  the year 1986.
- Inc year 170.

  2. That I have been employed by Works for 18 years as a regular employed in the field of molding technology.
- 3. That the following experiments were carried out by me or under my direct supervision and control.
- a) An aromatic polyamide (poly(phthalamide)) base resin was prepared by adding a filler material of bonic aluminum of an amount of 70% (by mass) thereto.
- b) A liquid crystal polyester base rests was prepared by adding a filler of fibrous potassium titanate at an amount of 50% (by mass) thereto.
- c) A polyether ether ketone was prepared by adding a filler of boric aluminum at an amount of 20% (by mass) thereso.

A sample of each base resin produced above was treated by mittegen plasma, oxygen plasma and argon plasma and a copper metal layer was deposited on the base resins using the procedures set forth in the present application at the section titled "Examples" on pages 29-30.

The adhesion between the base resin and the deposited copper was then measured and the results of the experiments are shown in Table 1.

Table 1

Base resin	Filler		Nitrogen	Oxygon	Argon
	Meterial	Configuration	plasma	plasma	1.04N/mm
Aromatic	boric	70%	1.IN/mm	0.77N/mm	1,0414,0414
polyanide	aluminum			, ,	
(bola(bpspsjsmigs))	(dismeter				
Christ Catronian .	0.5-1.0µm,				
	length 10-		}		
	30µm)			0.007/	0.37:N/mm
Laquid crystal	fibrous	50%	0.55N/mm	0.25N/mm	0-2 /// 271177
polyester	potassium	,			
pury date.	titenete	ļ			
1	(diameter			}	,
•	0.3-0.6µm,	}	Ĺ		1
•	length 10-		•	)	•
. •	20µm)				<del></del>
Polyether other	boric	20%	0.56N/mm	•	
ketoze	aluminum			1	•
Perome	(diameter		<b>\</b> '		
·. '	0.5-1.0µm,	}			. '
	length 10-			·	
· • • • • • • • • • • • • • • • • • • •	30µm)	}			

4. The results of the experiments set forth in Table 1 demonstrate for each different type of base resin containing a different amount of filler material in a different amount within the range of the present claims a higher adhesion for deposited metal to base resin treated by nitrogen plasma over deposited metal to a base resin treated by oxygen plasma or argon plasma from a range of approximately 6% greater adhesion up to 220% greater adhesion. Therefore, it is clear that nitrogen plasma treatment of a base resin companing filler material

according to Claim 1 produces superior adhesion between the base resin and deposited metal, as compared to adhesion between a base resin with filler material treated by oxygen plasma or argon plasma.

- 5. The undersigned petitioner declares further that all statements made herein of his own knowledge are true and that all statements made on information and belief are believed to be true; and further that these statements were made with the knowledge that willful false statements and the like so made are punishable by fine or imprisonment, or both, under Section 1001 of Title 18 of the United States Code and that such willful false statements may jeopardize the validity of this application or any patent issuing therefrom.
  - 6. Further deponent saith not

Signature NAKATA, Kimiaki

Qct. 27. 2004 Date

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Neoto IKEGAWA et al

GROUP ART UNIT: 1773

SERIAL NO: 09/871,896

: EXAMINER: Nikolas UHLIR

FILED: JUNE 4, 2001

FOR: LAMINATE

#### DECLARATION UNDER 37 C.F.R. 61.132

COMMISSIONER FOR PATENTS ALEXANDRIA, VIRGINIA 22913

SIR:

Now comes KeNDO, Naoyuki who deposes and states;

- 1. That I am a graduate of <u>University</u> and received a <u>Marter</u> degree in the year 1994.
- 2. That I have been employed by Electric for 10 years as a regular employed in the field of molding technology.
- 3. That the following experiments were carried out by me or under my direct expervision and control.
- a) An aromatic polyamide (poly(phthalamide)) base resin was prepared by adding a filler material of borde aluminum of an amount of 70% (by mass) thereto.
- b) A liquid crystal polyester base resin was prepared by adding a filler of filmous potassium titanate at an amount of 50% (by mass) thereto.
- c) A polyether other kotone was prepared by adding a filler of both aluminum at an amount of 20% (by mass) thereto.

A sample of each base resin produced above was treated by nitrogen plasma, oxygen plasma and argon plasma and a copper metal layer was deposited on the base resins using the procedures set forth in the present application at the section titled "Examples" on pages 29-30.

The adhesion between the base resin and the deposited copper was then measured and the results of the experiments are shown in Table 1.

Table 1

Base resin	Fi	ller	Nitrogen	Oxygen .	, blasms
DHZE 16971	Material	Configuration	plasma	plasma	1,04N/mm
Aromatic polyamide (poly(phthalamide))	boric eluminum (diameter 0.5-1.0µm, length 10- 30µm)	70%	1.1N/mm	0.77N/mm	
Liquid crystal polyester	fibrous potassium titanate (diamater 0.3-0.6µm. length 10- 20µm)	50%	0.55N/mm	0.25N/mm	0.37N/mm
Polyether ether ketone	boric . aluminum (diameter 0.5-1.0µm, length 10- 30µm)	20%	0.56N/mm		

4. The results of the experiments set forth in Table 1 demonstrate for each different type of base resin containing a different amount of filler material in a different amount within the range of the present claims a higher adhesion for deposited metal to base resin treated by nitrogen plasma over deposited metal to a base resin treated by oxygen plasma or argon plasma from a range of approximately 6% greater adhesion up to 220% greater adhesion. Therefore, it is clear that nitrogen plasma treatment of a base resin containing filler material.

according to Claim 1 produces superior adhesion between the base resin and deposited metal, as compared to adhesion between a base resin with filler material treated by oxygen plasma or argon plasma.

- 5. The undersigned petitioner declares further that all statements made herein of his own knowledge are true and that all statements made on information and bolief are believed to be true; and further that these statements were made with the knowledge that willful false statements and the like so made are punishable by fine or imprisonment, or both, under Section 1001 of Title 18 of the United States Code and that such willful false statements may jeopardize the validity of this application or any patent issuing therefrom.
  - 6. Further deponent saith not

Naoyuti Kondo Signatura kondo, Naoyuki

Oct. 27, 2004

Date

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